Superfolder Green Fluorescent Protein (sfGFP)

Orange Nanobody Enhancer

Green:

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Start by asking the right questions:

- 1) What do we need to learn?
- 2) Do we have individual reliable absorbance spectra?
- 3) What concentrations do we need to measure?
- 4) How do we get reliable molar extinction coefficients?
- 5) Are there any hypochromic/hyperchromic shifts upon binding?
- 6) What wavelengths need to be acquired?
- 7) How fast and long do we need to spin?
- 8) How do you interpret the molar mass of complexes? Need to know the molar masses of the individual species and their vbar values

Start by asking the right questions:

What do we need to learn?

- What do our controls look like by themselves in the AUC?
- **What are their molar extinction coefficients?**
- What are their binding strengths (k_D) ?
- **What are their binding stoichiometries?**
- **What are the molar masses of each species? (expected: free and complexed reactants)**

Start by asking the right questions:

Do we have individual, reliable absorbance spectra?

- Measure dilution series for each protein individually
- Determine the intrinsic molar extinction spectrum
- Overlay the two spectra to find regions of overlap and difference

Start by asking the right questions:

What concentrations do we need to measure?

- Make sure to stay within the dynamic range of the detector $(0.1-1.2 \text{ OD})$
- Decide if you are in the range of the k_D or much above or below
- This decides what type of experiment you can perform
- Use orthogonal information from ITC and MST to get ballpark k_D values
- Make sure you have enough protein
- Consider a titration experiment measuring multiple MW-AUC experiments

Start by asking the right questions:

1) **How do we get reliable molar extinction coefficients?**

- \cdot Rely on a protein that has a single, well known chromophore, with a calibrated molar extinction coefficient at one wavelength (e.g., aromatic amino acid side chains, a heme, or a fluorescent chromophore, and use that value to normalize the absorbance curve.
- Calibrate the unknown molecule by forming a complex and measuring the concentration and mass of the complex to decide on stoichiometry

Start by asking the right questions:

Are there any hypochromic/hyperchromic shifts upon binding?

- These are regions that change the shape or amplitude of a wavelength range upon binding
- Perform a titration experiment and normalize for concentration
- Overlay plots and visually inspect the sameness/difference for different ratios of ligands

Start by asking the right questions:

What wavelengths need to be acquired?

- Try to deconvolute a mixture of the two proteins to find the regions
- Find regions without hypochromic/hyperchromic shifts in the spectrum
- If there are shifts, determine their magnitude and determine in
- Minimize the wavelengths needed
- Use the most distinct wavelength in spectrum
- Maximize the angle between the spectral vectors using this formula:

$$
\theta = \cos^{-1}\left[\frac{u \cdot v}{\|u\| \cdot \|v\|}\right]
$$

Start by asking the right questions:

How fast and long do we need to spin?

- Optimize number of scans for each wavelength against resolution
- \bullet Faster sedimentation gives better resolution, but reduces # of scans before pelleting
- Simulate with the UltraScan software to get best combination
- Molar mass determination also requires good diffusion signal, which is enhanced in slower speed.
- Decide on the best compromise

Start by asking the right questions:

How do you interpret the molar mass of complexes? Need to know the molar masses of the individual species and their vbar values.

- Calculate the weight average partial specific volume of the proteins for a range of stoichiometries (using the solution tool in UltraScan)
- Plot the expected molar masses from s and D values for the complex when using the expected molar ratios vbar values
- Which molar mass/stoichiometry gives a matching molar mass when adding the individual molar masses? This is the key.