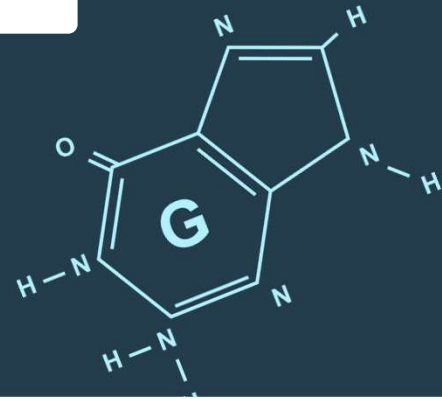
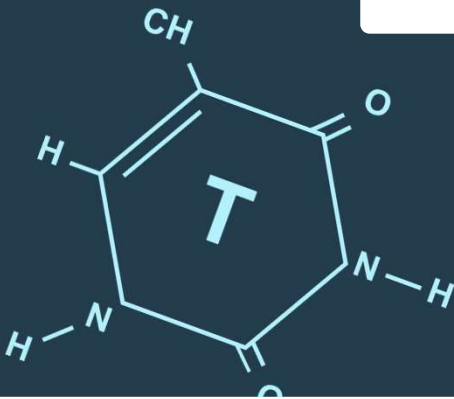


# Analytical Ultracentrifugation (AUC): An Overview of the Application of Fluorescence and Absorbance AUC to the Study of Biological Macromolecules

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## INTRO TO AUC

- **Definition:** A biophysical technique to study biological macromolecules in solution
- **Purpose:** Provides insights into size, shape, and interactions of molecules
- **Importance:** Widely used in biochemistry and molecular biology research.
- **Price of machines** are \$300,000-\$500,000

# BASICS OF AUC

- **Principle:** Measures sedimentation rates of molecules in a centrifugal field.
- **Components:** Rotor, sample cells, temperature, optical detection systems.
- **Advantages:** Direct measurement in solution state, wide size range coverage. Does not destroy the sample

Lamm equation:

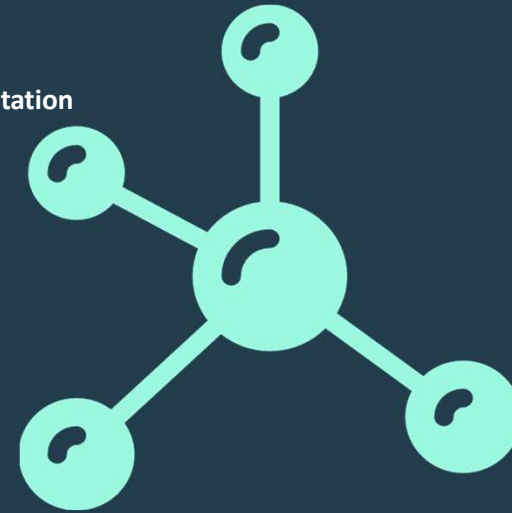
$$\frac{\partial c}{\partial t} = D \left[ \frac{\partial^2 c}{\partial r^2} + \frac{1}{r} \frac{\partial c}{\partial r} \right] - s\omega^2 \left[ r \frac{\partial c}{\partial r} + 2c \right]$$

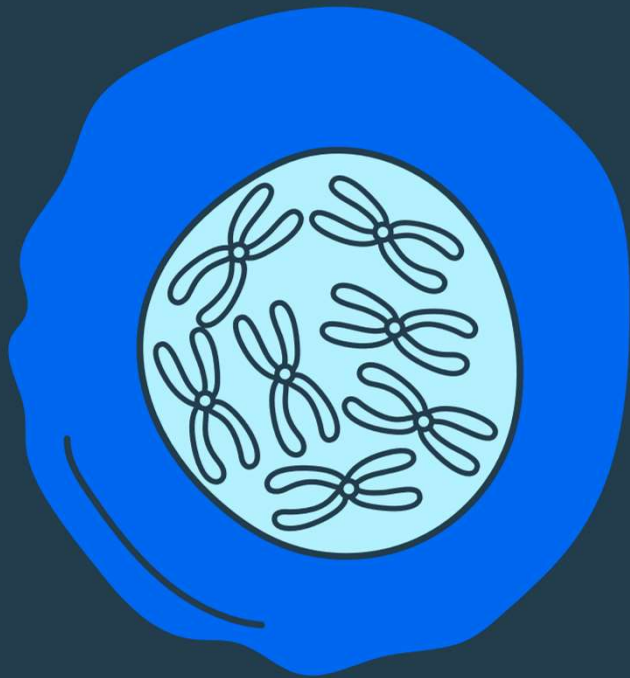
D = Diffusion coefficient  
c = particle concentration  
R = radius of cell  
t = time  
s = sedimentation coefficient

Svedberg equation:

$$\frac{M(1 - \bar{v}\rho)}{N_A f} = \frac{v}{\omega^2 r} \equiv s$$

S = sedimentation coefficient  
V = solute velocity  
R = radial distance from the axis of rotation  
 $\omega$  = angular (rotor) velocity  
M = molecular weight



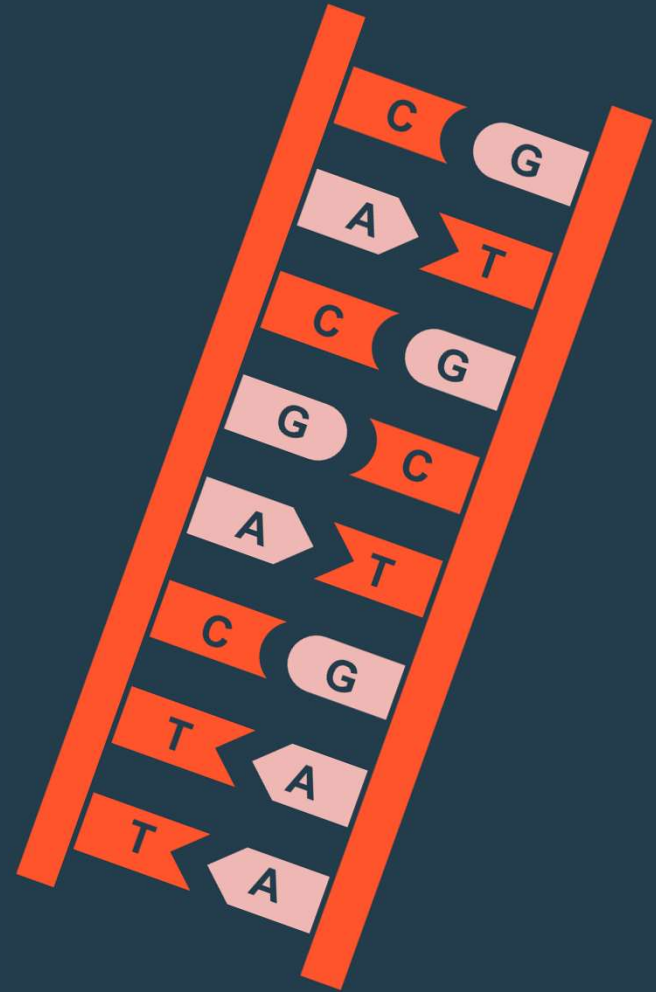


# TYPES OF AUC EXPERIMENTS

- **Sedimentation Velocity (SV):**  
Analyzes particle size and shape.
- **Sedimentation Equilibrium (SE):**  
Determines molecular weight and stoichiometry.
- **Fluorescence Detection:** Enhances sensitivity for low-concentration samples.

# APPLICATIONS OF AUC

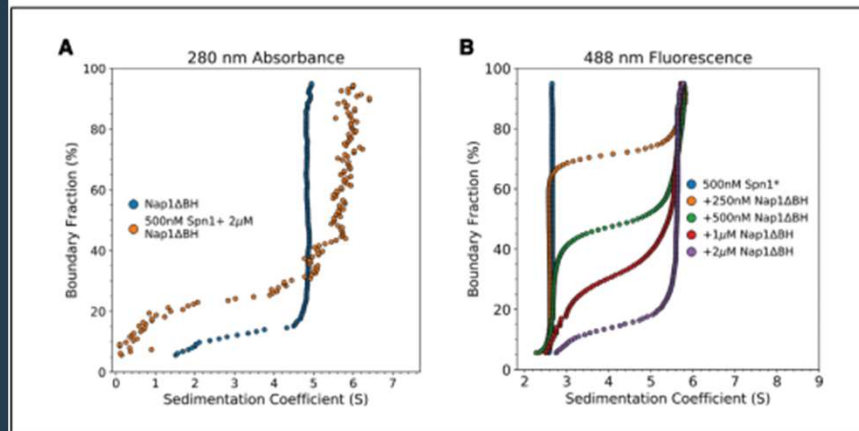
- **Protein-Protein Interactions:** Studying complex formation and stoichiometry.
- **Nucleic Acid Analysis:** Characterizing DNA-protein complexes.
- **Drug Development:** Assessing binding affinities and kinetics.



**Table 3** 2DSA-IT Results for Spn1 in Buffers With or Without 0.1% CHAPS Detergent

	Spn1	Spn1 (0.1% CHAPS)
$S_{20,W}$	2.57 (0.34) <sup>a</sup>	2.56 (0.36)
$f/f_0$	1.99 (0.56)	2.00 (1.14)
M (Da)	53,107 (31,332)	52,501 (31,574)

<sup>a</sup>Standard deviations are in parentheses. Peaks were integrated to include all S values observed by vHW.



**Figure 3** Fluorescence detection simplifies the sedimentation profile of a heterogeneous interacting system. (A) G(s) distributions from a 280-nm absorbance SV-AUC experiment. The majority of a sample of Nap1 Δ βH (blue) at 10 μM sediments homogeneously, whereas addition of 2 μM Nap1 Δ βH to 500 nM Spn1 (orange) results in a heterogeneous distribution of at least three states. (B) G(s) distributions from an SV-AUC-FDS experiment with Alexa488-labeled Spn1 and varying Nap1 Δ βH concentration. Binding results in a shift in fluorescence signal from ~2.6 S to ~5.6 S. The bound complex is more clearly resolved by the FDS than the equivalent sample monitored by absorbance.

**Table 4** Results from 2DSA-IT Analysis of Spn1-Nap1 Δ βH Samples from Case Study 1

	Spn1 (49.13 kDa)	+250 nM Nap1 Δ βH	+500 nM Nap1 Δ βH	+1 μM Nap1 Δ βH	+2 μM Nap1 Δ βH
$S_{20,W}$	2.58 (0.0108) <sup>a</sup>	Solute 1 (70%) = 2.62 (0.55) Solute 2 (18.5%) = 5.89 (n/a)	Solute 1 (43.8%) = 2.63 (0.542) Solute 2 (39.5%) = 5.59 (0.364)	Solute 1 (32.3%) = 2.605 (0.418) Solute 2 (60.6%) = 5.47 (0.434)	Solute 1 (12.9%) = 2.42 (0.586) Solute 2 (78.4%) = 5.63 (0.093)
$f/f_0$	2.01 (0.0523)	Solute 1 (70%) = 2.14 (0.71) Solute 2 (18.5%) = 2.58 (n/a)	Solute 1 (43.8%) = 2.38 (1.21) Solute 2 (39.5%) = 2.07 (0.5)	Solute 1 (32.3%) = 2.1 (1.9) Solute 2 (60.6%) = 2.18 (0.46)	Solute 1 (12.9%) = 2.17 (0.975) Solute 2 (78.4%) = 1.77 (0.463)
M (Da)	52,640 (2,216)	Solute 1 (70%) = 61,460 (28,731) Solute 2 (18.5%) = 262,188 (n/a)	Solute 1 (43.8%) = 67,772 (31,064) Solute 2 (39.5%) = 180,270 (76,092)	Solute 1 (32.3%) = 58,973 (67,599) Solute 2 (60.6%) = 188,070 (73,271)	Solute 1 (12.9%) = 61,076 (47,392) Solute 2 (78.4%) = 141,480 (51,794)

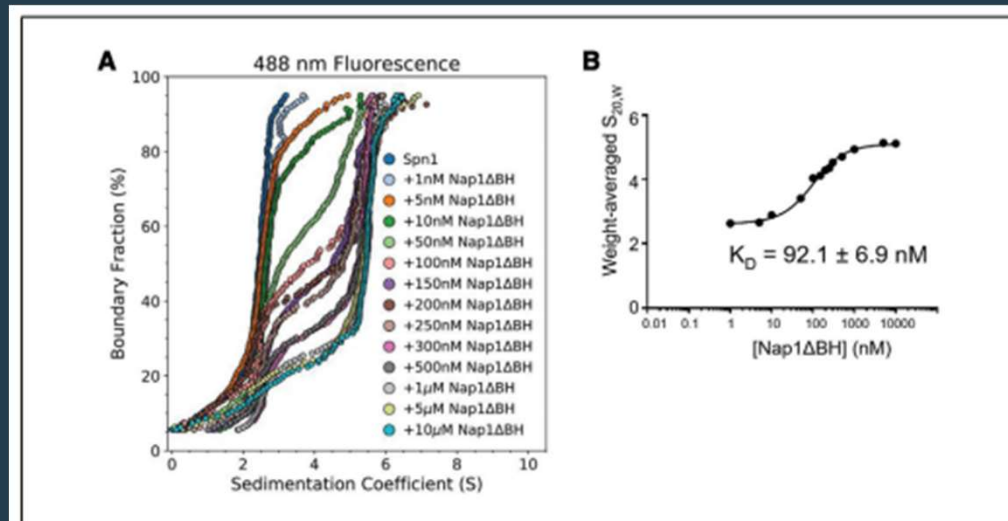
<sup>a</sup>Measured solute percentages and fitted parameter standard deviations are in parentheses. Peaks were integrated to include all S values observed by vHW.

## CASE STUDY 1: DETERMINING SIZE, SHAPE, AND STOICHIOMETRY

- **Example:** Analyzing histone chaperones Spn1 and Nap1 interaction.
- **Method:** Using absorbance and fluorescence AUC for size and stoichiometry determination.
- **Results:** Understanding the composition and binding sites of the protein complex. Gives much cleaner data
- **Theoretical MW: 148.18 kDa**
- **Experimental MW 141.48 kDa**

## CASE STUDY 2: MEASURING BINDING AFFINITIES

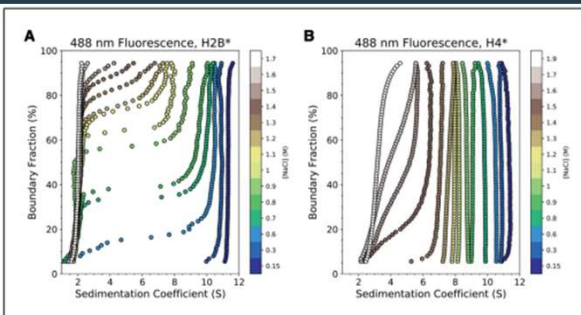
- **Example:** quantify the binding affinity between two biological macromolecules, Spn1 and Nap1 $\beta$ H, which was challenging to measure using other methods due to low signal changes upon binding.
- **Method:** Utilizing FDS to analyze high-affinity binding at nanomolar to micromolar
- **Results:** The experiment yielded a  $K_D$  of approximately 92.1 nM, indicating a strong binding affinity between Spn1 and Nap1 $\beta$ H.



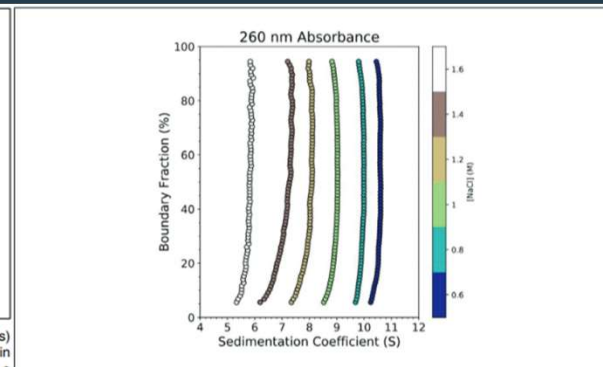
**Figure 5** Binding affinity quantitation by SV-AUC-FDS. **(A)** G(s) distributions from an SV-AUC-FDS experiment with Alexa488-labeled Spn1 (10 nM) and varying Nap1 $\Delta\beta$ H concentration. Binding results in a shift in fluorescence signal from  $\sim 2.6$  S to  $\sim 5.6$  S. **(B)** Weight-averaged  $S_{20,w}$  values of the single-experiment results in Figure 2A, plotted as a function of Nap1 $\Delta\beta$ H concentration and fit with GraphPad Prism's quadratic binding equation.



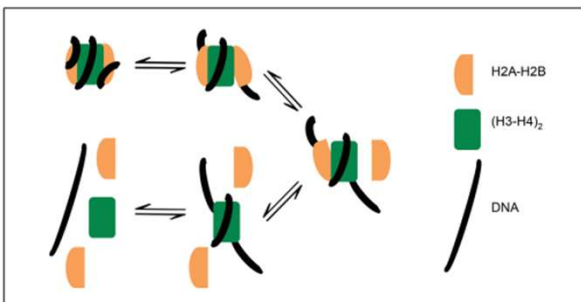
## CASE STUDY 3: INVESTIGATION OF MACROMOLECULAR ASSEMBLIES



**Figure 6** Salt-induced unfolding of eukaryotic nucleosomes monitored by SV-AUC-FDS. (A) G(s) distributions from an SV-AUC-FDS experiment with Alexa488-labeled H2B (T112C) included in 147-bp *X. laevis* nucleosomes starting at ~0 M NaCl. Dissociation of H2A-H2B\* is observed as a function of ionic strength at  $\geq 0.6$  M NaCl. (B) G(s) distributions from an SV-AUC-FDS experiment with Alexa488-labeled H4 (E63C) included in 147-bp *X. laevis* nucleosomes starting at ~0 M NaCl. An increasing NaCl concentration results in a decrease in sedimentation between 0.15 and 1.3 M, with dissociation of H3-H4 from DNA at higher salt concentrations.



**Figure 9** Salt-induced unfolding of eukaryotic nucleosomes monitored by 260-nm absorbance AUC. The G(s) distributions are from an SV-AUC 260-nm absorbance experiment with unlabeled 147-bp *X. laevis* nucleosomes. Nucleosomes were diluted from ~0 M NaCl into the indicated ionic strengths, resulting in nearly homogenous shifts of the DNA-dominated sedimentation signal.



**Figure 7** Schematic of the ionic strength-induced nucleosome folding/unfolding process observed in Case Study 3. Folded nucleosomes (top left) first partially unwrap, followed by sequential loss of H2A-H2B dimers. After H2A-H2B dissociation, increasing ionic strength dissociates (H3-H4)<sub>2</sub> from DNA, resulting in a mixture of DNA, H2A-H2B, and (H3-H4)<sub>2</sub>. These events are fully reversible, as shown in Figure 8.

**Table 6** Frictional Ratios of Partially Unfolded Nucleosome Samples from Case Study 3

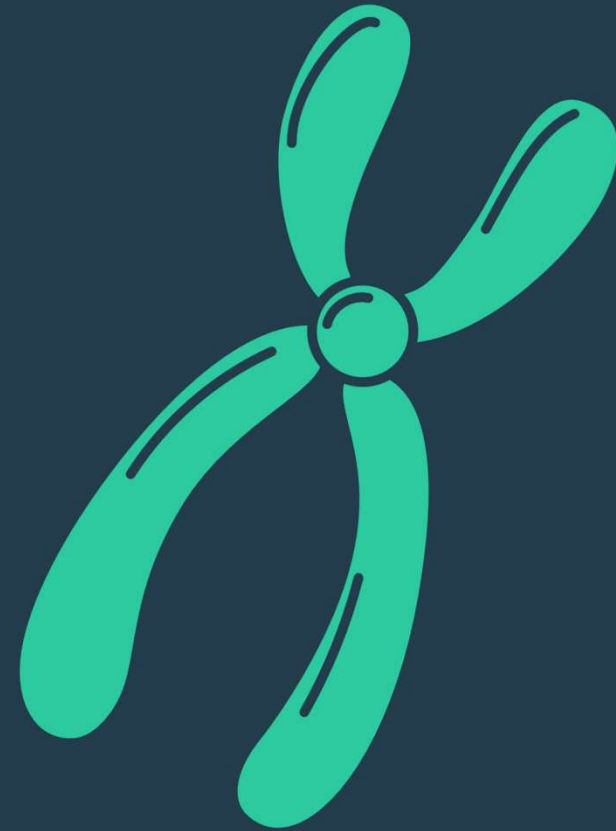
	0.15 M NaCl	0.3 M NaCl	0.6 M NaCl	0.7 M NaCl	0.8 M NaCl	0.9 M NaCl	1 M NaCl
$f/f_0$	1.47 (0.004) <sup>a</sup>	1.56 (0.005)	1.64 (0.006)	1.67 (0.008)	1.74 (0.005)	1.79 (0.061)	1.84 (0.06)

<sup>a</sup>Standard deviations are in parentheses. Values were extracted from 2DSA-IT analysis followed by genetic algorithm-Monte Carlo optimization. Peaks were integrated to include all S values observed by vHW. The observed trend is consistent across three replicates.

- **Example:** Characterization of protein complexes using AUC.
- **Method:** Applying AUC to determine assembly of complexes at various NaCl conc.
- **Results:** Understanding of complex structures and interactions. Number of complexes rough shapes and sizes by also utilizing various frictional ratios.

# DATA ANALYSIS IN AUC

- **Importance of User Curation:** Ensuring accurate interpretation of experimental data.
- **Model Fitting:** Evaluating model quality through visual representations and residuals.
- **Software Tools:** Utilizing UltraScan and SEDFIT for data analysis properly.
- **Warning:** User bias with limited understanding can falsely represent samples.





## **FUTURE DIRECTIONS IN AUC RESEARCH**

- **Advancements:** Continued development of software and instrumentation.
- **Opportunities:** Exploring AUC in diverse biological systems and complex interactions.
- **Recommendations:** Encouraging researchers to consider AUC in their projects, though with proper knowledge of how to apply it.



## CONCLUSION

- **Summary:** AUC is a valuable tool for studying macromolecular interactions.
- **Impact:** Enhances understanding of biological systems at the molecular level.
- **Limitation:** In particular AUC-FDS, requirement of fluorescent tags where 488nm is needed for excitation



## REFERENCE

Edwards, G. B., Muthurajan, U. M., Bowerman, S., & Luger, K. (2020). Analytical Ultracentrifugation (AUC): An Overview of the Application of Fluorescence and Absorbance AUC to the Study of Biological Macromolecules. *Current Protocols in Molecular Biology*, 133(1), e131.  
<https://doi.org/10.1002/cpmb.131>

**Thank you any  
question!**